

Across

2. Atom of an element that has a number of neutrons different from that of other atoms of the same element. (**isotope**)
8. Stimulated by the body for growth and repair and to make up enzymes. (**protein**)
9. Structure that contains the cell's genetic material and controls cell's activities. (**nucleus**)
11. Compound that produces hydroxide ions in solution. (**base**)
12. Smallest unit of most compounds. (**molecule**)
14. Compound that forms hydrogen ions in solution. (**acid**)
15. Process that changes one set of chemicals into another set of chemicals. (**chemical reaction**)
16. Element or compound that enters into a chemical reaction. (**reactant**)
23. Bond formed when one or more electrons are transferred from one atom to another. (**ionic bond**)
25. Large compound formed from combinations of many monomers. (**polymer**)
26. Link that holds together atoms in compounds. (**chemical bond**)
29. Bond formed by the sharing of electrons between atoms. (**covalent bond**)
32. Substance that is dissolved in a solvent to make a solution. (**solute**)
34. Element or compound produced by a chemical reaction. (**product**)
35. Attraction between molecules of different substances. (**adhesion**)
37. Substance consisting entirely of one type of atom. (**element**)
38. Macromolecules containing hydrogen, oxygen, nitrogen, carbon, and phosphorus. (**nucleic acid**)
40. Reactant of an end-enzyme-catalyzed reaction. (**substrate**)

Down

5. Material composed of two or more elements or compounds that are physically mixed together but not chemically combined. (**mixture**)
3. Measurement system used to indicate the concentration of hydrogen ions in solution. (**pH scale**)
4. Includes fats, oils, and waxes. (**lipid**)
5. Large macromolecule formed from monosaccharides. (**polysaccharide**)
6. Single sugar molecule. (**monosaccharide**)
7. Mixture of water and nondissolved materials. (**heterogeneous**)
10. Mixture of nucleic acids made up of a five-carbon sugar, phosphate group, and nitrogenous base. (**nucleotide**)
13. Energy needed to get a reaction started. (**activation energy**)
15. Weak acid or base that can react with strong acids or bases to help prevent sharp, sudden changes in pH. (**buffer**)
16. Substance formed by the chemical combination of two or more elements in definite proportions. (**compound**)
17. Single stranded nucleic acid that contains the sugar ribose. (**RNA**)
18. Small unit that can join together with other small units to form polymers. (**monomer**)
20. Negatively charged particles, located outside the atom's nucleus. (**electron**)
21. Compound with an amino group on one end and a carboxyl group on the other. (**amino acid**)
22. Mixture of two or more substances in which the molecules of the substances are evenly distributed. (**solution**)
24. Substance that speeds up the rate of a chemical reaction. (**catalyst**)
27. Major source of energy for the human body. (**carbohydrate**)
28. Attraction between molecules of the same substance. (**cohesion**)
30. Basic unit of matter. (**atom**)
33. Substance in which a solute is dissolved to form a solution. (**solvent**)
33. Protein that acts as a biological catalyst. (**enzyme**)
35. Atom that has a positive or negative charge. (**ion**)
36. Nucleic acid that contains the sugar deoxyribose. (**DNA**)

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